Significance of Previtamin D Chromatographic Resolution in the Accurate Determination of Vitamin D_3 by HPLC UV

David C. Woollard[1], Harvey E. Indyk[2]*, and Brendon D. Gill[2]

[1] Hill Laboratories, Private Bag 3205, Hamilton 3240, New Zealand

[2] Fonterra Co-operative Group Ltd, P.O. Box 7, Waitoa, New Zealand

* Corresponding author

Abstract

Conventional methods using HPLC with UV detection have used vitamin D2 as an internal standard with the expectation that this fully compensates for the heat-dependent equilibrium of vitamin D_3 with its previtamin. Previtamin D has a different spectral absorptivity from vitamin D and may be present in different proportions in samples and standards. Therefore, vitamin D_2 and vitamin D_3 and their previtamin forms must be chromatographically resolved to achieve accurate quantitation of total vitamin D. This study identified four chromatographic columns (ACE C_{18}, ACE C_{18} AR, Vydac 201 TP C_{18} and Polaris C_{18}-Ether) with adequate selectivity that should be applied for food testing and further confirmed that both parent vitamins isomerise at the same rate under thermal conditions.

Woollard, D.C.; Indyk, H.E.; Gill, B.D. (2019) Significance of Previtamin D Chromatographic Resolution in the Accurate Determination of Vitamin D_3 by HPLC UV. *Journal of Food Composition and Analysis* 79, 1–4. doi/org/10.1016/j.jfca.2019.02.010

1. Introduction

HPLC for vitamin D testing of foods has been used in routine analytical work for over three decades and although increasingly superseded by LC-MS/MS (Huang et al., 2009; Gilliland et al., 2013; Gill et al., 2015; AOAC 2016a), many laboratories continue to use UV-based methods (Vanhaelen-Fastré & Vanhaelen, 1981; Indyk & Woollard, 1985; Sertl & Molitor, 1985; O'Keefe & Murphy, 1988; Agarwal, 1989; Perales et al., 2005; Eitenmiller et al., 2008; AOAC 2016b). To achieve satisfactory analytical outcomes LC-UV methods rely on vitamin D_2 (ergocalciferol) as an internal standard to determine vitamin D_3 (cholecalciferol), and *vice versa* (Hagar et al., 1994; Salo-Väänänen et al., 2000), where a minimum chromatographic requirement is that vitamin D_2 and vitamin D_3 are resolved. The degree of separation is dependent on the intrinsic nature of the underlying silica and the surface bonding, and is traditionally accomplished using C_{18} or C_{30} columns (Indyk & Woollard, 1985; Sliva et al., 1992; Blanco et al., 2000; Staffas & Nyman, 2003; Huang et al., 2012). A C_{30} column has greater hydrophobicity and provides stronger retention of fat-soluble vitamins (Huang et al., 2009).

The temperature-dependent formation of previtamin D, as illustrated in Fig. 1, has been demonstrated by several researchers (Keverling Buisman et al., 1968; Mulder et al., 1971; de Vries et al., 1979). Endogenous levels of previtamin D_3 in a food product, in contrast to the absence of previtamin D_2 in the internal standard, will contribute to a small, but significant quantitative error due to different spectral absorptivities (Gill et al., 2019). Using vitamin D_2 as the internal standard therefore requires not only adequate chromatographic resolution of vitamin D_2 and vitamin D_3 but also chromatographic resolution of both previtamins. Optimum vitamin D_2–vitamin D_3 separation alone does not imply effective quantitation, as previtamin D_3 often co-elutes with vitamin D_2 and cannot be differentiated using UV detection, therefore potentially contributing to minor analytical bias.

Provided that saponification conditions are sufficient for vitamin D and previtamin D from both sample and internal standard to reach equilibrium, it is not necessary to quantitate previtamin D peaks in the analysis of total vitamin D_3 (sum of previtamin D_3 and vitamin D_3). However, few reported analytical methods accomplish this requirement, resulting in unequal proportions of previtamin D forms from sample and internal standard (Gill et al., 2019). Under such conditions it is important to ensure that the previtamin forms are chromatographically resolved from the parent vitamin forms.

The principle aim of this study therefore, was to evaluate a wide range of analytical chromatographic columns and to identify those with the capability to achieve baseline resolution of the four vitamin D forms.

2. Materials and methods

2.1. Apparatus

Analytical chromatography was performed on an Agilent HPLC system (Santa Clara, CA, USA), with G1322 A degasser, G1311 A quaternary pump, G1329B autosampler, G1316-A column compartment and G1315 A diode array detector, or on a Shimadzu Prominence HPLC system (Columbia, MD, USA), with LC-20AD pump and DGU-20 A degasser unit, SPD-M20 A diode array detector, CTO-20AC column oven and sample injection by a Shimadzu Nexera SIL-30AC autosampler at 10 °C.

Other equipment used were a three-decimal place balance for weighing samples (Ohaus Explorer Pro, Parsippany, NJ, USA) and four decimal place balances for standards (Sartorius, Göttingen, Germany).

2.2. Columns

Separations were performed on 16 reversed-phase columns in 5 µm fully porous and 2.6 µm superficially porous (core-shell) formats, to identify those with sufficient chromatographic separation of vitamin D_2, vitamin D_3, previtamin D_2 and previtamin D_3. Columns selected for further studies were Acclaim C_{30} (Thermo Fisher, Waltham, MA, USA), ACE C_{18} and ACE C_{18}-AR (Advanced Chromatography Technologies, Aberdeen, Scotland), Vydac 201 TP C_{18} (Hichrom, Reading, England), Microsorb 300-5 and Polaris C_{18}-Ether (Agilent, Santa Clara, CA, USA); all were 5 µm fully porous and 4.6 × 250 mm.

2.3. Reagents

Vitamin standards (≥ 98%), cholecalciferol (vitamin D_3) and ergocalciferol (vitamin D_2), were obtained from Sigma Aldrich (St. Louis, MO, USA). Methanol and acetonitrile of HPLC grade were obtained from Thermo Fisher Scientific (Waltham, MA, USA). Standards were prepared in acetonitrile and UV exposure minimised by using amber glassware under UV-filtered lighting.

2.4. Standards

Following preliminary temperature studies, all columns were operated at 30 °C under the same isocratic conditions with an acetonitrile mobile phase. The 5 µm columns were operated at 1.0 mL/min and the 2.6 µm core-shell columns at 0.7 mL/min, and under these conditions, standard solutions were monitored for changes in vitamin D and previtamin D. The primary monitoring wavelength used was 265 nm with spectral data in the range 200–350 nm acquired for peak identity and purity assessments. Previtamin D peak identifications were achieved by heating standard solutions for 1 h at 60 °C and monitoring proportional changes in peak areas.

2.5. Isomerisation experiments

Freshly prepared solutions of vitamin D_2 and vitamin D_3 standards were subjected to time trials at ambient temperature (18 ± 2 °C) for the appearance of their respective previtamins. At each time point, extracts were placed in a −18 °C freezer until ready for analysis. Peak identities were confirmed by the changes in area with time and by spectral scanning, in which previtamin D isomers have maxima at 262 nm, i.e. slightly lower than their parent calciferols. Peak areas were normalised using published extinction coefficients ($E^{1\%}$), with previtamin D_2 and previtamin D_3 peak areas multiplied by 2.28 (475/208) and 2.22 (485/218) respectively (Hanewald et al., 1968).

3. Results and discussion

Incomplete chromatographic resolution of the four calciferol vitamers will lead to compromised quantitation and an accurate estimate of total vitamin D_3 content can only be obtained when there is complete resolution of the previtamin D peaks. It is therefore of concern that the majority of C_{18} columns do not facilitate complete resolution of all four relevant vitamin D forms.

3.1. Method development and validation

Adequate vitamin D_2 and vitamin D_3 separation (Rs > 1.8) could be achieved with all 16 commercial C_{18} columns evaluated. However, the capability to also separate previtamin D from their respective vitamin D_2 and vitamin D_3 parents was limited to only four of the columns: ACE C_{18}, ACE C_{18}-AR, Agilent Polaris C_{18}-Ether and Vydac 201 TP. The column evaluations were all performed isocratically in acetonitrile without attempting to optimise with solvent changes and gradient elution. Even though some columns might not have been operating optimally, it was considered expedient to compare column performance under common conditions. If the four vitamin D chromatographic peaks were not resolved under these conditions, the column will not be fit for purpose for more complex food extracts.

Columns that achieved acceptable resolution of all four previtamin D and vitamin D peaks are illustrated in Fig. 2. The ACE C_{18} column had a separation capability that was superior to that of the ACE C_{18}-AR column; the latter column is designed for enhanced aromatic selectivity, indicating that the vitamin D steroidal structure predictably does not interact with the available aryl functionality. The equivalent ACE C_{18}-PFP column had a further reduction in resolution that was commensurate with its increased polarity, leading to co-elution of previtamin D_3 and vitamin D_2. The Polaris C_{18} Ether column, with an end-capped ether group to improve polar retention, assisted vitamin D separations presumably through its hydroxyl functionality. In addition, Polaris silica has 180 Å pores, i.e. larger than the 100 Å ACE silica chemistry. Speculation that pore size is important originates from the ability

of the Vydac 201 TP column, with 300 Å pores, to efficiently separate calciferols. However, the selectivity of the Vydac 201 TP column is different from those of both the ACE column and the Polaris C_{18}-Ether column, in that previtamin D_3 elutes between vitamin D_2 and vitamin D_3 (Fig. 2). Trials using an equivalent 300 Å C_{18} Microsorb 300-5 column exhibited similar tendencies to separate the calciferols, although with less efficiency than the Vydac column. Both 300 Å columns had significant peak tailing when operated in 100% acetonitrile, which could possibly be improved with a methanol:water gradient.

In contrast to these columns, it was noted that for the enhanced hydrophobic Acclaim C_{30} column, the previtamin forms eluted after their respective parent vitamins; nonetheless, this did not solve quantitation problems as previtamin D_2 co-eluted with vitamin D_3. This reversal of elution order had been noted by Huang et al. (2009) using a YMC S3 carotenoid C_{30} column. A core-shell pentafluorophenyl column showed separation in the same order as for the C_{30} columns (Wei et al., 2017).

3.2. Isomerisation

In its *cis*-configuration, the target measurand vitamin D_3, will readily interconvert with previtamin D_3, and the related interconversion also occurs with vitamin D_2. Measurable levels of previtamin D were absent in freshly prepared vitamin D_2 or vitamin D_3 standard solutions, but were rapidly formed dependent on the temperature of storage. For this reason, standards were kept at −18 °C for up to 6 months and under such conditions, the extent of isomerisation remained negligible. The vitamin D_2 and vitamin D_3 standards were removed from the freezer, divided into 10 vials and kept at ambient temperature for successive days and returned to the freezer at each time point to prevent further conversion. After 10 days, all standards were analysed simultaneously, where the results confirmed that vitamin D_2 and vitamin D_3 isomerised to their previtamin forms at the same rate (Fig. 3).

As expected, previtamin D peaks increased in area at approximately half the rate of the concomitant decrease in parent vitamin, due to their different spectral absorptivities. This supports the applicability of vitamin D_2 as a surrogate for vitamin D_3, compensating both for manipulative losses during difficult multi-stage sample preparations and for heat-induced isomerisation, and is consistent with previous literature (Hanewald et al., 1968). However, as discussed, this attribute is applicable only when both previtamin D forms are chromatographically resolved from each other and their parent forms under HPLC-UV conditions.

4. Conclusions

When HPLC-UV analysis is employed for the analysis of vitamin D_3, incorporating vitamin D_2 as an internal standard, it is important to fully separate the four vitamin D forms. In an acetonitrile mobile phase, only four of sixteen C_{18} columns were found to achieve this, with the Vydac 201 TP C_{18} column demonstrating a selectivity that was different from that of the other columns. Columns capable of adequate previtamin resolution in model standard solutions should be favoured for future vitamin D determinations in food matrices to avoid potential bias due to inadequate resolution of previtamin D forms. Also, it was confirmed that the rate and extent of thermal isomerisation of vitamin D_2 and vitamin D_3 to their previtamin forms were equivalent, supporting the utility of ergocalciferol as an internal standard.

Declarations of interest

None.

Acknowledgements

The authors acknowledge the support of Fonterra Cooperative Group Ltd. This research did not receive any specific grant from funding agencies in the public, commercial, or not-for-profit sectors.

References

Agarwal, V.K., 1989. Liquid chromatographic determination of vitamin D in infant formula. Journal of the *Association of Official Analytical Chemists* 72, 1007–1009.

AOAC, 2016a. Official Methods of Analysis of AOAC International 20th ed.: Method 2016.05. AOAC International, Gaithersburg, MD.

AOAC, 2016b. Official Methods of Analysis of AOAC International 20th ed.: Method 2002.05. AOAC International, Gaithersburg, MD.

Blanco, D., Fernández, M.P., Gutiérrez, M.D., 2000. Simultaneous determination of fat soluble vitamins and provitamins in dairy products by liquid chromatography with a narrow-bore column. *Analyst* 125, 427–431.

de Vries, E.J., Zeeman, J., Esser, R.J., Borsje, B., Mulder, F.J., 1979. Analysis of fat-soluble vitamins. XXIII. High performance liquid chromatographic assay for vitamin D in vitamin D_3 and multivitamin preparations. *Journal of the Association of Official Analytical Chemists* 62, 1285–1291.

Eitenmiller, R.R., Ye, L., Landen Jr, W.O., 2008. *Vitamin analysis for the health and food sciences*, 2nd ed. CRC Press, Boca Raton, FL, USA.

Gill, B.D.; Gilliland, D.L.; Indyk, H.E.; Wood, J.E.; Woollard, D.C. (2019) Significance of thermal isomerisation on the quantitation of total vitamin D_3 in foods. *Food Analytical Methods* 12, 998–106. DOI: 10.1007/s12161-019-01434-6

Gill, B.D., Zhu, X., Indyk, H.E., 2015. A rapid method for the determination of vitamin D_3 in milk and infant formula by liquid chromatography/tandem mass spectrometry. *Journal of AOAC International* 98, 431–435.

Gilliland, D.L., Black, C.K., Denison, J.E., Seipelt, C.T., Baugh, S., 2013. Simultaneous determination of vitamins D2 and D_3 by electrospray ionization LC/MS/MS in infant formula and adult nutritionals: First Action 2012.11. *Journal of AOAC International* 96, 1387–1395.

Hagar, A.F., Madsen, L., Wales Jr, L., Bradford Jr, H.B., 1994. Reversed-phase liquid chromatographic determination of vitamin D in milk. *Journal of AOAC International* 77, 1047–1051.

Hanewald, K.H., Mulder, F.J., Keuning, K.J., 1968. Thin-layer chromatographic assay of vitamin D in high-potency preparations. Analysis of fat-soluble vitamins IX. *Journal of Pharmaceutical Sciences* 57, 1308–1312.

Huang, M., LaLuzerne, P., Winters, D., Sullivan, D., 2009. Measurement of vitamin D in foods and nutritional supplements by liquid chromatography/tandem mass spectrometry. Journal of AOAC International 92, 1327–1335.

Huang, M., Winters, D., Sullivan, D., Dowell, D., 2012. Application of ultra-high-performance liquid chromatography/tandem mass spectrometry for the measurement of vitamin D in infant formula and adult/pediatric nutritional formula: First Action 2011.11. *Journal of AOAC International* 95, 319–321.

Indyk, H.E., Woollard, D.C., 1985. The determination of vitamin D in fortified milk powders and infant formulas by HPLC. *Journal of Micronutrient Analysis* 1, 121–141.

Keverling Buisman, J.A., Hanewald, K.H., Mulder, F.J., Roborgh, J.R., Keuning, K.J., 1968. Evaluation of the effect of isomerization on the chemical and biological assay of vitamin D. Analysis of fat-soluble vitamins X. *Journal of Pharmaceutical Sciences* 57, 1326–1329.

Mulder, F.J., de Vries, E.J., Borsje, B., 1971. Chemical analysis of vitamin D in concentrates and its problems. XII. Analysis of fat-soluble vitamins. *Journal of the Association of Official Analytical Chemists* 54, 1168–1174.

O'Keefe, S.F., Murphy, P.A., 1988. Rapid determination of vitamin D in fortified skim milk. *Journal of Chromatography* 445, 305–309.

Perales, S., Alegría, A., Barberá, R., Farré, R., 2005. Determination of vitamin D in dairy products by high performance liquid chromatography. *Food Science and Technology International* 11, 451–462.

Salo-Väänänen, P., Ollilainen, V., Mattila, P., Lehikoinen, K., Salmela-Mölsä, E., Piironen, V., 2000. Simultaneous HPLC analysis of fat-soluble vitamins in selected animal products after small-scale extraction. *Food Chemistry* 71, 535–543.

Sertl, D.C., Molitor, B.E., 1985. Liquid chromatographic determination of vitamin D in milk and infant formula. *Journal of the Association of Official Analytical Chemists* 68, 177–182.

Sliva, M.G., Green, A.E., Sanders, J.K., Euber, J.R., Saucerman, J.R., 1992. Reversed phase liquid chromatographic determination of vitamin D in infant formulas and enteral nutritionals. *Journal of AOAC International* 75, 566–571.

Staffas, A., Nyman, A., 2003. Determination of cholecalciferol (vitamin D3) in selected foods by liquid chromatography: NMKL collaborative study. *Journal of AOAC International* 86, 400–406.

Vanhaelen-Fastré, R., Vanhaelen, M., 1981. Separation and determination of the D vitamins by HPLC. In: Kautsky, M.P. (Ed.*), Steroid analysis by HPLC: recent application*s. Marcel Dekker, New York, NY, USA, pp. 173–251.

Wei, X., Trass, M., Misa, A., Krepich, S., Oriowicz, S., 2017. *Determination of Vitamin D2/ D_3 and Pre-D2/D_3 in Pet Food by LC/MS/MS*. Application Note TN-0111. Phenomenex, Torrance, CA, USA (Accessed June 2018). https://www.phenomenex.com/ViewDocument?id=determination+of+vitamin+d2_d3+and+pre-d2_d3+in +pet+food+by+lc_ms_ms

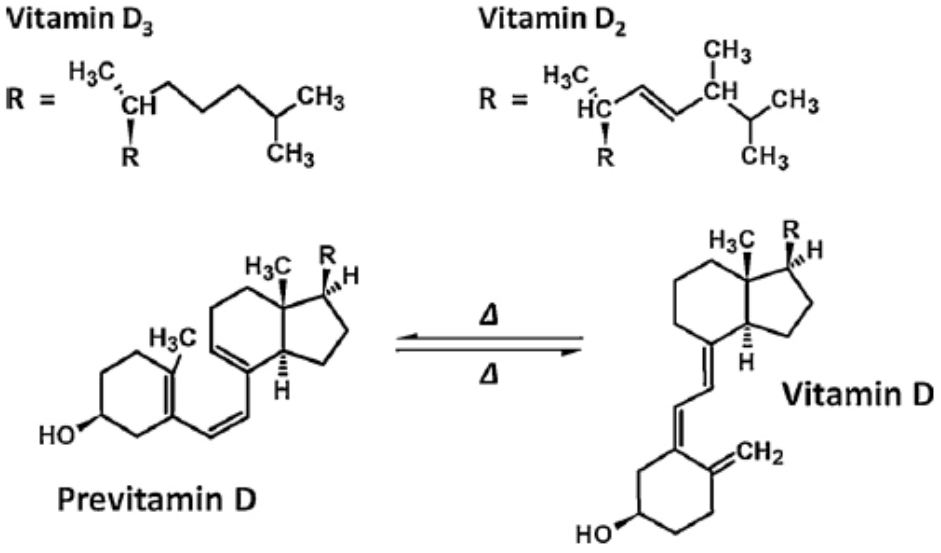

Fig. 1. Interconversion of previtamin D and vitamin D

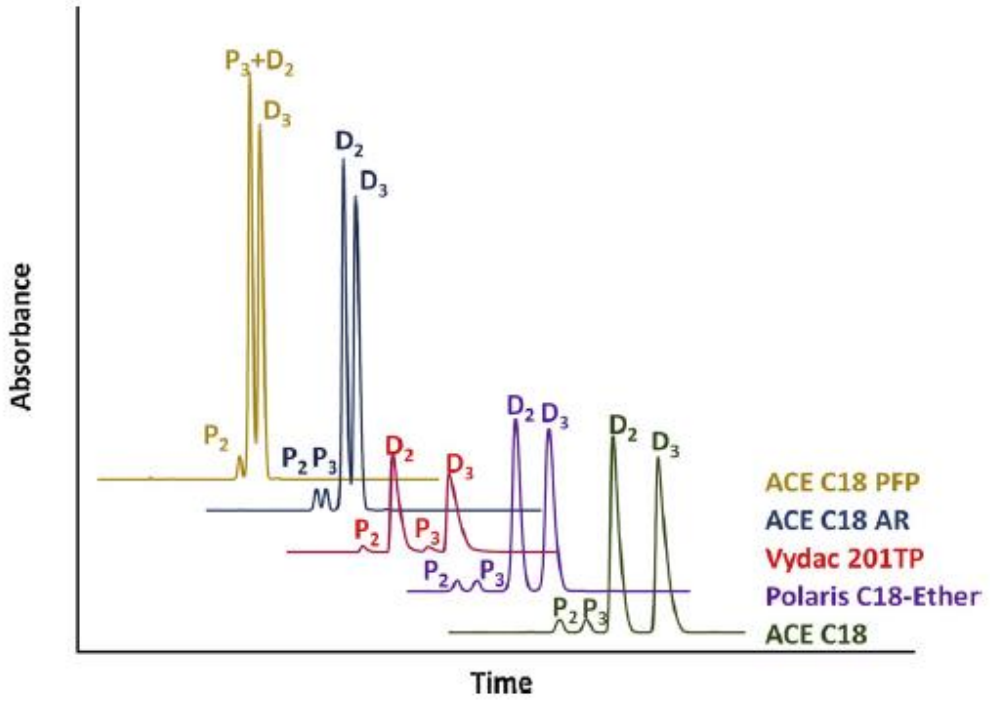

Fig. 2. Separation of vitamin D_2 and vitamin D_3 and their previtamin forms using selected columns (P=previtamin D; D=vitamin D)

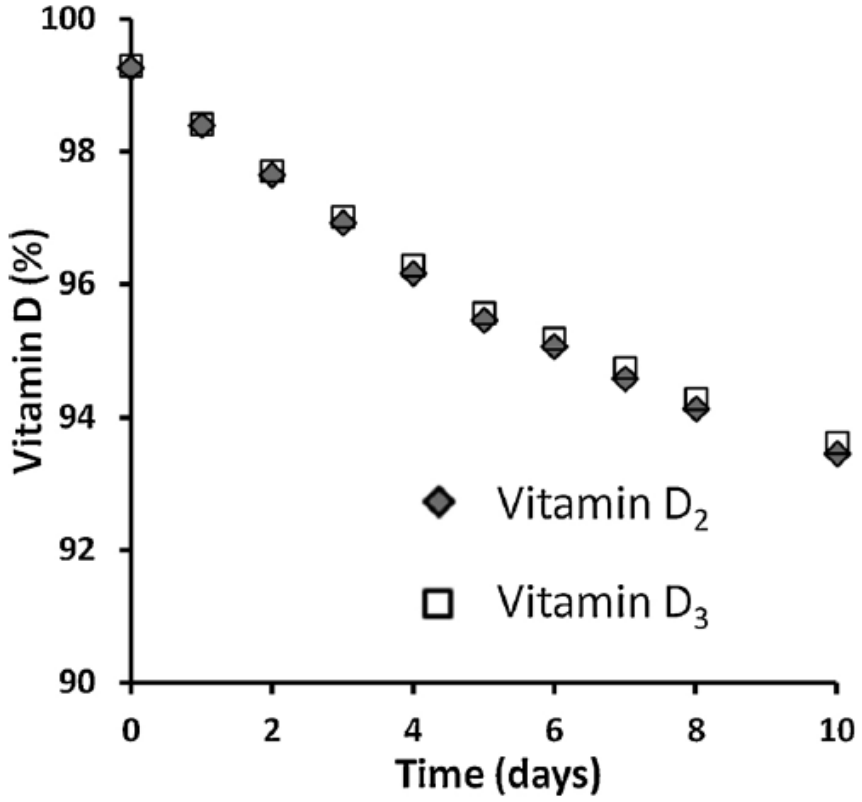

Fig. 3. Decrease in vitamin D_2 and vitamin D_3 as a result of thermal isomerisation at room temperature (18 ± 2 °C)