



Significance of Previtamin D Chromatographic Resolution in the Accurate Determination of Vitamin D₃ by HPLC UV

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Abstract

Conventional methods using HPLC with UV detection have used vitamin D₂ as an internal standard with the expectation that this fully compensates for the heat-dependent equilibrium of vitamin D₃ with its previtamin. Previtamin D has a different spectral absorptivity from vitamin D and may be present in different proportions in samples and standards. Therefore, vitamin D₂ and vitamin D₃ and their previtamin forms must be chromatographically resolved to achieve accurate quantitation of total vitamin D. This study identified four chromatographic columns (ACE C₁₈, ACE C₁₈ AR, Vydac 201 TP C₁₈ and Polaris C₁₈-Ether) with adequate selectivity that should be applied for food testing and further confirmed that both parent vitamins isomerise at the same rate under thermal conditions.

1. Introduction

HPLC for vitamin D testing of foods has been used in routine analytical work for over three decades and although increasingly superseded by LC-MS/MS (Huang et al., 2009; Gilliland et al., 2013; Gill et al., 2015; AOAC 2016a), many laboratories continue to use UV-based methods (Vanhaelen-Fastré & Vanhaelen, 1981; Indyk & Woollard, 1985; Sertl & Molitor, 1985; O’Keefe & Murphy, 1988; Agarwal, 1989; Perales et al., 2005; Eitenmiller et al., 2008; AOAC 2016b). To achieve satisfactory analytical outcomes LC-UV methods rely on vitamin D₂ (ergocalciferol) as an internal standard to determine vitamin D₃ (cholecalciferol), and *vice versa* (Hagar et al., 1994; Salo-Väänänen et al., 2000), where a minimum chromatographic requirement is that vitamin D₂ and vitamin D₃ are resolved. The degree of separation is dependent on the intrinsic nature of the underlying silica and the surface bonding, and is traditionally accomplished using C₁₈ or C₃₀ columns (Indyk & Woollard, 1985; Sliva et al., 1992; Blanco et al., 2000; Staffas & Nyman, 2003; Huang et al., 2012). A C₃₀ column has greater hydrophobicity and provides stronger retention of fat-soluble vitamins (Huang et al., 2009).

The temperature-dependent formation of previtamin D, as illustrated in Fig. 1, has been demonstrated by several researchers (Keverling Buisman et al., 1968; Mulder et al., 1971; de Vries et al., 1979). Endogenous levels of previtamin D₃ in a food product, in contrast to the absence of previtamin D₂ in the internal standard, will contribute to a small, but significant quantitative error due to different spectral absorptivities (Gill et al., 2019). Using vitamin D₂ as the internal standard therefore requires not only adequate chromatographic resolution of vitamin D₂ and vitamin D₃ but also chromatographic resolution of both previtamins. Optimum vitamin D₂–vitamin D₃ separation alone does not imply effective quantitation, as previtamin D₃ often co-elutes with vitamin D₂ and cannot be differentiated using UV detection, therefore potentially contributing to minor analytical bias.

Provided that saponification conditions are sufficient for vitamin D and previtamin D from both sample and internal standard to reach equilibrium, it is not necessary to quantitate previtamin D peaks in the analysis of total vitamin D₃ (sum of previtamin D₃ and vitamin D₃). However, few reported analytical methods accomplish this requirement, resulting in unequal proportions of previtamin D forms from sample and internal standard (Gill et al., 2019). Under such conditions it is important to ensure that the previtamin forms are chromatographically resolved from the parent vitamin forms.

The principle aim of this study therefore, was to evaluate a wide range of analytical chromatographic columns and to identify those with the capability to achieve baseline resolution of the four vitamin D forms.

2. Materials and methods

2.1. Apparatus

Analytical chromatography was performed on an Agilent HPLC system (Santa Clara, CA, USA), with G1322 A degasser, G1311 A quaternary pump, G1329B autosampler, G1316-A column compartment and G1315 A diode array detector, or on a Shimadzu Prominence HPLC system (Columbia, MD, USA), with LC-20AD pump and DGU-20 A degasser unit, SPD-M20 A diode array detector, CTO-20AC column oven and sample injection by a Shimadzu Nexera SIL-30AC autosampler at 10 °C.

Other equipment used were a three-decimal place balance for weighing samples (Ohaus Explorer Pro, Parsippany, NJ, USA) and four decimal place balances for standards (Sartorius, Göttingen, Germany).

2.2. Columns

Separations were performed on 16 reversed-phase columns in 5 µm fully porous and 2.6 µm superficially porous (core-shell) formats, to identify those with sufficient chromatographic separation of vitamin D₂, vitamin D₃, previtamin D₂ and previtamin D₃. Columns selected for further studies were Acclaim C₃₀ (Thermo Fisher, Waltham, MA, USA), ACE C₁₈ and ACE C₁₈-AR (Advanced Chromatography Technologies, Aberdeen, Scotland), Vydac 201 TP C₁₈ (Hichrom, Reading, England), Microsorb 300-5 and Polaris C₁₈-Ether (Agilent, Santa Clara, CA, USA); all were 5 µm fully porous and 4.6 × 250 mm.

2.3. Reagents

Vitamin standards (≥ 98%), cholecalciferol (vitamin D₃) and ergocalciferol (vitamin D₂), were obtained from Sigma Aldrich (St. Louis, MO, USA). Methanol and acetonitrile of HPLC grade were obtained from Thermo Fisher Scientific (Waltham, MA, USA). Standards were prepared in acetonitrile and UV exposure minimised by using amber glassware under UV-filtered lighting.

2.4. Standards

Following preliminary temperature studies, all columns were operated at 30 °C under the same isocratic conditions with an acetonitrile mobile phase. The 5 µm columns were operated at 1.0 mL/min and the 2.6 µm core-shell columns at 0.7 mL/min, and under these conditions, standard solutions were monitored for changes in vitamin D and previtamin D. The primary monitoring wavelength used was 265 nm with spectral data in the range 200–350 nm acquired for peak identity and purity assessments. Previtamin D peak identifications were achieved by heating standard solutions for 1 h at 60 °C and monitoring proportional changes in peak areas.

2.5. Isomerisation experiments

Freshly prepared solutions of vitamin D₂ and vitamin D₃ standards were subjected to time trials at ambient temperature (18 ± 2 °C) for the appearance of their respective previtamins. At each time point, extracts were placed in a -18 °C freezer until ready for analysis. Peak identities were confirmed by the changes in area with time and by spectral scanning, in which previtamin D isomers have maxima at 262 nm, i.e. slightly lower than their parent calciferols. Peak areas were normalised using published extinction coefficients ($E^{1\%}$), with previtamin D₂ and previtamin D₃ peak areas multiplied by 2.28 (475/208) and 2.22 (485/218) respectively (Hanewald et al., 1968).

3. Results and discussion

Incomplete chromatographic resolution of the four calciferol vitamers will lead to compromised quantitation and an accurate estimate of total vitamin D₃ content can only be obtained when there is complete resolution of the previtamin D peaks. It is therefore of concern that the majority of C₁₈ columns do not facilitate complete resolution of all four relevant vitamin D forms.

3.1. Method development and validation

Adequate vitamin D₂ and vitamin D₃ separation ($R_s > 1.8$) could be achieved with all 16 commercial C₁₈ columns evaluated. However, the capability to also separate previtamin D from their respective vitamin D₂ and vitamin D₃ parents was limited to only four of the columns: ACE C₁₈, ACE C₁₈-AR, Agilent Polaris C₁₈-Ether and Vydac 201 TP. The column evaluations were all performed isocratically in acetonitrile without attempting to optimise with solvent changes and gradient elution. Even though some columns might not have been operating optimally, it was considered expedient to compare column performance under common conditions. If the four vitamin D chromatographic peaks were not resolved under these conditions, the column will not be fit for purpose for more complex food extracts.

Columns that achieved acceptable resolution of all four previtamin D and vitamin D peaks are illustrated in Fig. 2. The ACE C₁₈ column had a separation capability that was superior to that of the ACE C₁₈-AR column; the latter column is designed for enhanced aromatic selectivity, indicating that the vitamin D steroidal structure predictably does not interact with the available aryl functionality. The equivalent ACE C₁₈-PFP column had a further reduction in resolution that was commensurate with its increased polarity, leading to co-elution of previtamin D₃ and vitamin D₂. The Polaris C₁₈ Ether column, with an end-capped ether group to improve polar retention, assisted vitamin D separations presumably through its hydroxyl functionality. In addition, Polaris silica has 180 Å pores, i.e. larger than the 100 Å ACE silica chemistry. Speculation that pore size is important originates from the ability

of the Vydac 201 TP column, with 300 Å pores, to efficiently separate calciferols. However, the selectivity of the Vydac 201 TP column is different from those of both the ACE column and the Polaris C₁₈-Ether column, in that previtamin D₃ elutes between vitamin D₂ and vitamin D₃ (Fig. 2). Trials using an equivalent 300 Å C₁₈ Microsorb 300-5 column exhibited similar tendencies to separate the calciferols, although with less efficiency than the Vydac column. Both 300 Å columns had significant peak tailing when operated in 100% acetonitrile, which could possibly be improved with a methanol:water gradient.

In contrast to these columns, it was noted that for the enhanced hydrophobic Acclaim C₃₀ column, the previtamin forms eluted after their respective parent vitamins; nonetheless, this did not solve quantitation problems as previtamin D₂ co-eluted with vitamin D₃. This reversal of elution order had been noted by Huang et al. (2009) using a YMC S3 carotenoid C₃₀ column. A core-shell pentafluorophenyl column showed separation in the same order as for the C₃₀ columns (Wei et al., 2017).

3.2. Isomerisation

In its *cis*-configuration, the target measurand vitamin D₃, will readily interconvert with previtamin D₃, and the related interconversion also occurs with vitamin D₂. Measurable levels of previtamin D were absent in freshly prepared vitamin D₂ or vitamin D₃ standard solutions, but were rapidly formed dependent on the temperature of storage. For this reason, standards were kept at –18 °C for up to 6 months and under such conditions, the extent of isomerisation remained negligible. The vitamin D₂ and vitamin D₃ standards were removed from the freezer, divided into 10 vials and kept at ambient temperature for successive days and returned to the freezer at each time point to prevent further conversion. After 10 days, all standards were analysed simultaneously, where the results confirmed that vitamin D₂ and vitamin D₃ isomerised to their previtamin forms at the same rate (Fig. 3).

As expected, previtamin D peaks increased in area at approximately half the rate of the concomitant decrease in parent vitamin, due to their different spectral absorptivities. This supports the applicability of vitamin D₂ as a surrogate for vitamin D₃, compensating both for manipulative losses during difficult multi-stage sample preparations and for heat-induced isomerisation, and is consistent with previous literature (Hanewald et al., 1968). However, as discussed, this attribute is applicable only when both previtamin D forms are chromatographically resolved from each other and their parent forms under HPLC-UV conditions.

4. Conclusions

When HPLC-UV analysis is employed for the analysis of vitamin D₃, incorporating vitamin D₂ as an internal standard, it is important to fully separate the four vitamin D forms. In an acetonitrile mobile phase, only four of sixteen C₁₈ columns were found to achieve this, with the Vydac 201 TP C₁₈ column demonstrating a selectivity that was different from that of the other columns. Columns capable of adequate previtamin resolution in model standard solutions should be favoured for future vitamin D determinations in food matrices to avoid potential bias due to inadequate resolution of previtamin D forms. Also, it was confirmed that the rate and extent of thermal isomerisation of vitamin D₂ and vitamin D₃ to their previtamin forms were equivalent, supporting the utility of ergocalciferol as an internal standard.

Declarations of interest

None.

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References

- Agarwal, V.K., 1989. Liquid chromatographic determination of vitamin D in infant formula. *Journal of the Association of Official Analytical Chemists* 72, 1007–1009.
- AOAC, 2016a. Official Methods of Analysis of AOAC International 20th ed.: Method 2016.05. AOAC International, Gaithersburg, MD.
- AOAC, 2016b. Official Methods of Analysis of AOAC International 20th ed.: Method 2002.05. AOAC International, Gaithersburg, MD.
- Blanco, D., Fernández, M.P., Gutiérrez, M.D., 2000. Simultaneous determination of fat soluble vitamins and provitamins in dairy products by liquid chromatography with a narrow-bore column. *Analyst* 125, 427–431.
- de Vries, E.J., Zeeman, J., Esser, R.J., Borsje, B., Mulder, F.J., 1979. Analysis of fat-soluble vitamins. XXIII. High performance liquid chromatographic assay for vitamin D in vitamin D₃ and multivitamin preparations. *Journal of the Association of Official Analytical Chemists* 62, 1285–1291.

- Eitenmiller, R.R., Ye, L., Landen Jr, W.O., 2008. *Vitamin analysis for the health and food sciences*, 2nd ed. CRC Press, Boca Raton, FL, USA.
- Gill, B.D.; Gilliland, D.L.; Indyk, H.E.; Wood, J.E.; Woollard, D.C. (2019) Significance of thermal isomerisation on the quantitation of total vitamin D₃ in foods. *Food Analytical Methods* 12, 998–1006. DOI: 10.1007/s12161-019-01434-6
- Gill, B.D., Zhu, X., Indyk, H.E., 2015. A rapid method for the determination of vitamin D₃ in milk and infant formula by liquid chromatography/tandem mass spectrometry. *Journal of AOAC International* 98, 431–435.
- Gilliland, D.L., Black, C.K., Denison, J.E., Seipelt, C.T., Baugh, S., 2013. Simultaneous determination of vitamins D₂ and D₃ by electrospray ionization LC/MS/MS in infant formula and adult nutritionals: First Action 2012.11. *Journal of AOAC International* 96, 1387–1395.
- Hagar, A.F., Madsen, L., Wales Jr, L., Bradford Jr, H.B., 1994. Reversed-phase liquid chromatographic determination of vitamin D in milk. *Journal of AOAC International* 77, 1047–1051.
- Hanewald, K.H., Mulder, F.J., Keuning, K.J., 1968. Thin-layer chromatographic assay of vitamin D in high-potency preparations. Analysis of fat-soluble vitamins IX. *Journal of Pharmaceutical Sciences* 57, 1308–1312.
- Huang, M., LaLuzerne, P., Winters, D., Sullivan, D., 2009. Measurement of vitamin D in foods and nutritional supplements by liquid chromatography/tandem mass spectrometry. *Journal of AOAC International* 92, 1327–1335.
- Huang, M., Winters, D., Sullivan, D., Dowell, D., 2012. Application of ultra-high-performance liquid chromatography/tandem mass spectrometry for the measurement of vitamin D in infant formula and adult/pediatric nutritional formula: First Action 2011.11. *Journal of AOAC International* 95, 319–321.
- Indyk, H.E., Woollard, D.C., 1985. The determination of vitamin D in fortified milk powders and infant formulas by HPLC. *Journal of Micronutrient Analysis* 1, 121–141.
- Keverling Buisman, J.A., Hanewald, K.H., Mulder, F.J., Roborgh, J.R., Keuning, K.J., 1968. Evaluation of the effect of isomerization on the chemical and biological assay of vitamin D. Analysis of fat-soluble vitamins X. *Journal of Pharmaceutical Sciences* 57, 1326–1329.
- Mulder, F.J., de Vries, E.J., Borsje, B., 1971. Chemical analysis of vitamin D in concentrates and its problems. XII. Analysis of fat-soluble vitamins. *Journal of the Association of Official Analytical Chemists* 54, 1168–1174.

- O’Keefe, S.F., Murphy, P.A., 1988. Rapid determination of vitamin D in fortified skim milk. *Journal of Chromatography* 445, 305–309.
- Perales, S., Alegría, A., Barberá, R., Farré, R., 2005. Determination of vitamin D in dairy products by high performance liquid chromatography. *Food Science and Technology International* 11, 451–462.
- Salo-Väänänen, P., Ollilainen, V., Mattila, P., Lehtikainen, K., Salmela-Mölsä, E., Piironen, V., 2000. Simultaneous HPLC analysis of fat-soluble vitamins in selected animal products after small-scale extraction. *Food Chemistry* 71, 535–543.
- Sertl, D.C., Molitor, B.E., 1985. Liquid chromatographic determination of vitamin D in milk and infant formula. *Journal of the Association of Official Analytical Chemists* 68, 177–182.
- Sliva, M.G., Green, A.E., Sanders, J.K., Euber, J.R., Saucerman, J.R., 1992. Reversed phase liquid chromatographic determination of vitamin D in infant formulas and enteral nutritionals. *Journal of AOAC International* 75, 566–571.
- Staffas, A., Nyman, A., 2003. Determination of cholecalciferol (vitamin D₃) in selected foods by liquid chromatography: NMKL collaborative study. *Journal of AOAC International* 86, 400–406.
- Vanhaelen-Fastré, R., Vanhaelen, M., 1981. Separation and determination of the D vitamins by HPLC. In: Kautsky, M.P. (Ed.), *Steroid analysis by HPLC: recent applications*. Marcel Dekker, New York, NY, USA, pp. 173–251.
- Wei, X., Trass, M., Misa, A., Krepich, S., Oriowicz, S., 2017. *Determination of Vitamin D₂/D₃ and Pre-D₂/D₃ in Pet Food by LC/MS/MS*. Application Note TN-0111. Phenomenex, Torrance, CA, USA (Accessed June 2018).
- https://www.phenomenex.com/ViewDocument?id=determination+of+vitamin+d2_d3+and+pre-d2_d3+in+pet+food+by+lc_ms_ms

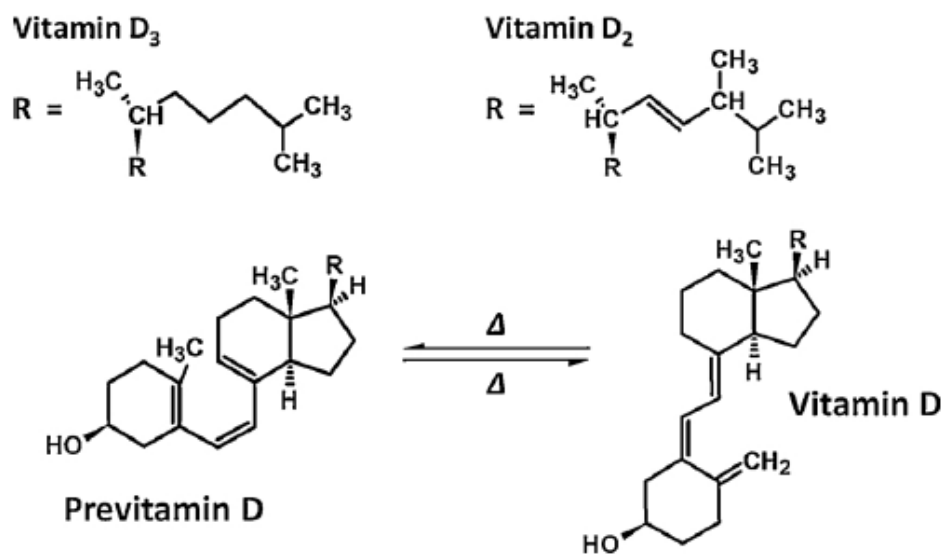


Fig. 1. Interconversion of previtamin D and vitamin D

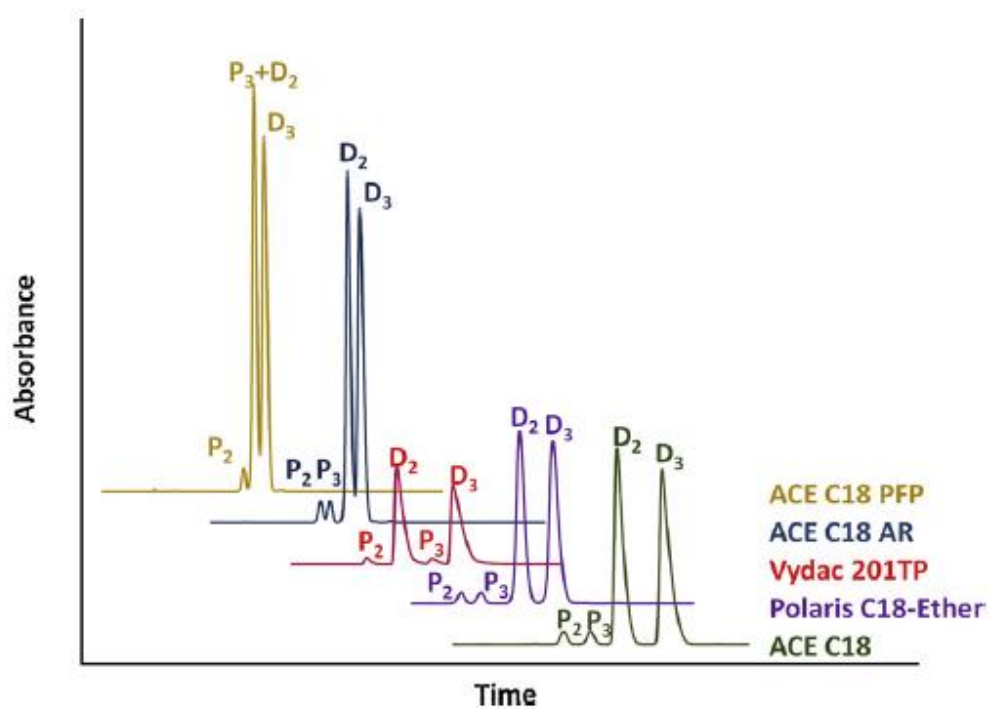


Fig. 2. Separation of vitamin D₂ and vitamin D₃ and their previtamin forms using selected columns (P=previtamin D; D=vitamin D)

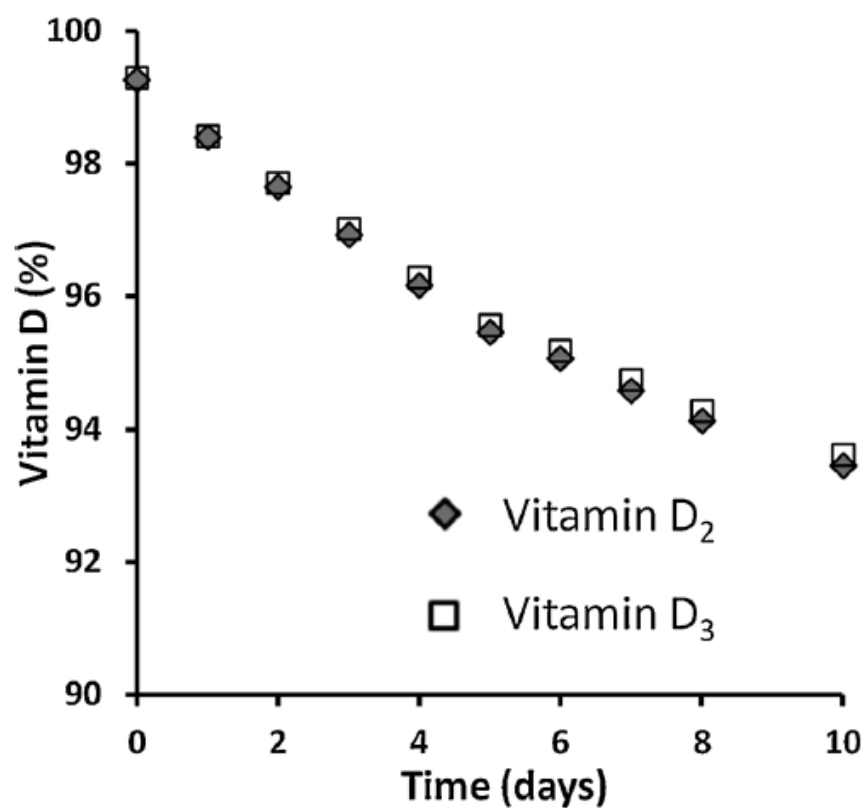


Fig. 3. Decrease in vitamin D₂ and vitamin D₃ as a result of thermal isomerisation at room temperature (18 ± 2 °C)